# CHEMISTRY IN AQUEOUS AND NON-AQUEOUS SOLUTIONS

For B.Sc Chemistry(Part-II) Inorganic chemistry Paper-III L.ecture-06

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### **Contents**

- Physical properties of Liquid SO<sub>2</sub>
- Application/uses of Liquid SO<sub>2</sub>
- Inorganic reactions in liq. SO<sub>2</sub>
- Inorganic reactions of liq. SO<sub>2</sub> Solutions of metals in liq NH<sub>3</sub>

### Physical properties of Liquid SO<sub>2</sub>

- □ Colour: Colourless as a gas or liquid
- Odour : Intense, pungent, irritating
- Flammability: Not flammable, not explosive in either the gaseous or liquid state
- Molecular Formula: SO<sub>2</sub>
- Molecular Weight :64.063
- Boiling Point: -10.1°C (14°F)

#### **Application:**

- > Liquid sulphur dioxide (SO2) is a versatile chemical
- > Used both in **liquid** form or as a source of gaseous SO2
- > Used in the food industry as a preservative
- > Used the pulp and paper industry
- Used in mining industry
- Liq SO2 is also be used as a refrigerant, heat transfer fluid and selective solvent

#### Liquid sulphur dioxide (SO<sub>2</sub>)

#### Acid -Base reaction

Liq So2 undergoes auto ionization as

$$2 \text{ SO}_2 \longrightarrow \text{SO}^{+2} + \text{SO}_3^{-2}$$

Acid -Base reaction or neutralization reaction can be rerpresented as

$$SOCl_2 + CS_2O_3 \longrightarrow 2CSCl + 2SO_2$$

Acid Base