

For B.Sc –III Chemistry(H/S) Physical Chemistry Paper-V Lecture-01

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Spectroscopy is the study of the interaction between matter and electromagnetic radiation.

spectroscopy originated through the study of visible light dispersed according to its wavelength, by a prism.

It uses the techniques of radiation to obtain information on the structure and properties of matter.

## What is Spectroscopy



## **IMPORTANCE OF SPECTROSCOPY**

- It helps in determining the composition, temperature, density, and motion of an object.
- It helps to identify the atoms and molecules in the object.
- The red shift or blue shift (Doppler Effect) in a spectral line tells how fast the object is receding from Earth or coming toward it.

# **TYPES OF ABSORPTION SPECTROSCOPY**

The most common types of waves measured by absorption spectroscopy are Infrared Visible Ultraviolet (UV) Atomic X-ray Each spectrophotometer works using the same techniques. Spectrophotometers are categorized according to the type of wave being measured.

## **Basis principles of different types of absorption**

#### Principle of absorption

- Absorption spectroscopy refers to spectroscopic techniques that measure the absorption of radiation, as a function of frequency or wavelength, due to its interaction with a sample.
- The sample absorbs energy, i.e. photons, from the radiating field.
- The technique uses basically the principle that free atoms (gas) generated in an atomizer can absorb radiation at specific frequency.
- Atomic-absorption spectroscopy quantifies the absorption of ground state atoms in the gaseous state.

#### How is absorption measured?

There are many different approaches for measuring absorption spectra.

- When a beam of light generated from the sample passes through it then detect the intensity of the radiation
- The energy that is then transmitted is used to calculate the absorption.
- Wavelength of maximum absorption ( $\lambda$ Max) the extent to which a sample absorbs light depends upon the wavelength of light.
- The wavelength at which a subtance shows maximum absorbance is called absorption Maximum or lamda max.

Note: Value of  $\lambda$  max is important for several reasons.

## **Basic Instrumentation of Spectrophotometer**



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